

PATRIK LOMBER


U.S. **UTILITY** Patent Application

9101 O.I.P.E.

PATENT DATE

SCANNED

Q.A.

APPLICATION NO. 09/574836	CONT/PRIOR F	CLASS 428	SUBCLASS 35.7	ART UNIT 1772	EXAMINER 
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TITLE	APPLICANTS
1. The effect of the concentration of the solution on the rate of reaction.	Mr. J. H. Smith
2. The effect of the temperature of the solution on the rate of reaction.	Mr. J. H. Smith
3. The effect of the surface area of the solid reactant on the rate of reaction.	Mr. J. H. Smith
4. The effect of the catalyst on the rate of reaction.	Mr. J. H. Smith
5. The effect of the concentration of the solution on the equilibrium constant.	Mr. J. H. Smith
6. The effect of the temperature of the solution on the equilibrium constant.	Mr. J. H. Smith
7. The effect of the pressure on the equilibrium constant.	Mr. J. H. Smith
8. The effect of the volume of the gas on the equilibrium constant.	Mr. J. H. Smith
9. The effect of the concentration of the solution on the pH value.	Mr. J. H. Smith
10. The effect of the temperature of the solution on the pH value.	Mr. J. H. Smith
11. The effect of the concentration of the solution on the conductivity.	Mr. J. H. Smith
12. The effect of the temperature of the solution on the conductivity.	Mr. J. H. Smith
13. The effect of the concentration of the solution on the viscosity.	Mr. J. H. Smith
14. The effect of the temperature of the solution on the viscosity.	Mr. J. H. Smith
15. The effect of the concentration of the solution on the refractive index.	Mr. J. H. Smith
16. The effect of the temperature of the solution on the refractive index.	Mr. J. H. Smith
17. The effect of the concentration of the solution on the density.	Mr. J. H. Smith
18. The effect of the temperature of the solution on the density.	Mr. J. H. Smith
19. The effect of the concentration of the solution on the boiling point.	Mr. J. H. Smith
20. The effect of the temperature of the solution on the boiling point.	Mr. J. H. Smith
21. The effect of the concentration of the solution on the freezing point.	Mr. J. H. Smith
22. The effect of the temperature of the solution on the freezing point.	Mr. J. H. Smith
23. The effect of the concentration of the solution on the melting point.	Mr. J. H. Smith
24. The effect of the temperature of the solution on the melting point.	Mr. J. H. Smith
25. The effect of the concentration of the solution on the solubility.	Mr. J. H. Smith
26. The effect of the temperature of the solution on the solubility.	Mr. J. H. Smith
27. The effect of the concentration of the solution on the diffusion coefficient.	Mr. J. H. Smith
28. The effect of the temperature of the solution on the diffusion coefficient.	Mr. J. H. Smith
29. The effect of the concentration of the solution on the osmotic pressure.	Mr. J. H. Smith
30. The effect of the temperature of the solution on the osmotic pressure.	Mr. J. H. Smith
31. The effect of the concentration of the solution on the vapor pressure.	Mr. J. H. Smith
32. The effect of the temperature of the solution on the vapor pressure.	Mr. J. H. Smith
33. The effect of the concentration of the solution on the heat of fusion.	Mr. J. H. Smith
34. The effect of the temperature of the solution on the heat of fusion.	Mr. J. H. Smith
35. The effect of the concentration of the solution on the heat of vaporization.	Mr. J. H. Smith
36. The effect of the temperature of the solution on the heat of vaporization.	Mr. J. H. Smith
37. The effect of the concentration of the solution on the heat of combustion.	Mr. J. H. Smith
38. The effect of the temperature of the solution on the heat of combustion.	Mr. J. H. Smith
39. The effect of the concentration of the solution on the heat of formation.	Mr. J. H. Smith
40. The effect of the temperature of the solution on the heat of formation.	Mr. J. H. Smith
41. The effect of the concentration of the solution on the heat of solution.	Mr. J. H. Smith
42. The effect of the temperature of the solution on the heat of solution.	Mr. J. H. Smith
43. The effect of the concentration of the solution on the heat of dilution.	Mr. J. H. Smith
44. The effect of the temperature of the solution on the heat of dilution.	Mr. J. H. Smith
45. The effect of the concentration of the solution on the heat of crystallization.	Mr. J. H. Smith
46. The effect of the temperature of the solution on the heat of crystallization.	Mr. J. H. Smith
47. The effect of the concentration of the solution on the heat of polymerization.	Mr. J. H. Smith
48. The effect of the temperature of the solution on the heat of polymerization.	Mr. J. H. Smith
49. The effect of the concentration of the solution on the heat of decomposition.	Mr. J. H. Smith
50. The effect of the temperature of the solution on the heat of decomposition.	Mr. J. H. Smith

Jean-Claude Engelere

Egar

Resealable packaging the container of which includes a tearable welding layer, and its manufacture

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